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Hydrogen sulfide (H₂S) Content Assay Kit

Note: It is necessary to predict 2-3 large difference samples before the formal determination.

Operation Equipment: Spectrophotometer

Catalog Number: BC2050

Size: 50T/48S

Components:

Reagent	Size	Storage
Extract solution I	Solution 60 mL×1	2-8°C
Extract solution II	Solution 10 mL×1	2-8°C
Reagent I	Solution 20 mL×1	2-8°C
Reagent II	Solution 20 mL×1	2-8°C

Product Description:

Hydrogen sulfide (H_2S) is a new type of gaseous signal molecule. It is a neurotransmitter that exists in the brain. The physiological concentration of H_2S has an important regulatory effect on the long-term enhancement of the hippocampus of the nervous system. It also plays an important pathophysiological effect on the process of spontaneous hypertension, hemorrhagic shock and liver cirrhosis.

H₂S can react with N, N-dimethyl-p-phenylenediamine and ferric ammonium sulfate to form methylene blue. Methylene blue has a maximum absorption peak at 665nm. The H₂S content can be calculated by measuring the absorbance value.

Reagents and Equipment Required but Not Provided:

Spectrophotometer, desk centrifuge, pipette, 1mL glass cuvette, mortar/homogenizer, ice and distilled water.

Procedure

I. Sample preparation:

- Bacteria or cells: collecting bacteria or cells into the centrifuge tube, centrifugation and discard supernatant. Suggest add 1 mL of Extract solution I to 5 million of bacteria or cells. Use ultrasonication to splitting bacteria and cell (placed on ice, ultrasonic power 200W, ultrasonic 3 seconds, interval 7 seconds, total time 3 minutes). Centrifuge at 12000 ×g for 10 minutes at 4°C to remove insoluble materials. Then add 0.15 mL Extract solution II to 0.8 mL supernatant. Centrifuge at 12000 ×g for 10 minutes at 4°C to remove insoluble materials. Take the supernatant on ice for test.
- 2. Tissue: add 1 mL of Extract solution I into 0.1 g of tissue and fully grind on ice. Centrifuge at 12000 ×g for 10minutes at 4°C to remove insoluble materials. Then add 0.15 mL Extract solution II to 0.8 mL supernatant. Centrifuge at 12000 ×g for 10 minutes at 4°C to remove insoluble materials. Take the supernatant on ice for test.

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3. Serum (plasma) or other liquid samples: add 1 mL of Extract solution I into 1 mL of serum (plasma). Centrifuge at 12000 ×g for 10minutes at 4°C to remove insoluble materials. Then add 0.15 mL Extract solution II to 0.8 mL supernatant. Centrifuge at 12000 ×g for 10 minutes at 4°C to remove insoluble materials. Take the supernatant on ice for test.

II. Determination procedure:

1. Preheat spectrophotometer for 30min, adjust wavelength to 665 nm, set zero with distilled water.

2. Determination:

Test tube	Blank tube
250	-
-	250
375	375
375	375
	250 - 375

Mix well. React at room temperature for 10 minutes. Measure the absorbance at 665 nm, record as A_T , A_B . Calculate the $\Delta A=A_T-A_B$. Blank tube only need to be test one or two times.

III. Calculations:

Take the concentration of standard solution(nmol/mL) as x-axis, and the corresponding ΔA is y-axis. Then the linear regression equation y = 0.0026x-0.0268, R²=0.9973 is obtained. Bring ΔA into the equation to get x (nmol/mL).

1. Protein concentration:

H₂S content (nmol/mg prot) = $x \times V_S \div (V_S \times Cpr) = x \div Cpr$

2. Sample weight:

H₂S content (nmol/g weight) = $x \times (V_{SP} + V_{EX2}) \div (W \times V_{SP} \div V_{EX1}) = 1.1875 \times x \div W$

3. Cell amount:

H₂S content (nmol/10⁴ cell) = $x \times (V_{SP} + V_{EX2}) \div$ (cells × $V_{SP} \div V_{EX1}$)=1.1875×x÷cells

4. Serum (plasma) sample:

H₂S content (nmol/mL) = $x \times (V_{SP} + V_{EX2}) \div [(V_L \times V_{SP} \div (V_{EX1} + V_L)] = 13.0625 \times x$

Vs: Sample volume in reaction, 0.25 mL;

V_{SP}: Supernatant volume in Extraction, 0.8 mL;

V_{EX1}: Extraction solution I volume, 1 mL;

V_{EX2}: Extraction solution II volume, 0.15 mL;

Cpr: Sample protein concentration, mg/mL;

W: Sample weight, g;

cells: Total number of bacteria and cells, 104;

V_L: Liquid sample volume, 0.1 mL.

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Note:

1. If the ΔA is lower, it is recommended to increase the sample size before determination; If $\Delta A > 0.6$, it is recommended to dilute the sample before determination. The calculation formula should be multiplied by the corresponding dilution factor.

Examples:

1. Take 0.1g of mouse liver to follow the determination procedure to operate. Determination with 1mL glass cuvette, and calculate $\Delta A = A_T - A_B = 0.090 - 0.026 = 0.064$. The calculated content is as follows:

H₂S content (nmol/g weight) = $x \times (V_{SP} + V_{EX2}) \div (W \times V_{SP} \div V_{EX1}) = 414.71$ nmol/g weight.

2. Take 0.1g of ginkgo leaf plum to follow the determination procedure to operate. Determination with 1mL glass cuvette, and calculate $\Delta A = A_T - A_B = 0.081 - 0.026 = 0.055$. The calculated content is as follows:

H₂S content (nmol/g weight) = $x \times (V_{SP} + V_{EX2}) \div (W \times V_{SP} \div V_{EX1}) = 373.61$ nmol/g weight.

Related products:

BC1170/ BC1175	Reduced Glutathione (GSH) Assay Kit	
BC1180/ BC1185	Oxidized Glutathione (GSSG) Assay Kit	
BC1190/ BC1195	Glutathione Peroxidase (GPX) Assay Kit	
BC1150/ BC1155	Oxidized Thioredoxin Reductase (TrxR) Assay Kit	
BC1470/BC1475	Nitric oxide (NO) Assay Kit	



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