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Acid xylanase (ACX) Activity Assay Kit

Note: The reagents of this product have changed, please operate in strict accordance with the instructions.

Detection equipment: Spectrophotometer **Cat No:** BC2600

Size: 50T/24S

Components:

Buffer Fluid: Liquid 75 mL×1, store at 2-8°C.

Reagent I: Liquid 20 mL×1, store at 2-8°C.

Reagent II: Liquid 25 mL×1, store at 2-8°C.

Reagent III: Liquid 8 mL×1, store at 2-8°C.

Standard: Powder×1, 10mg xylose. Before use, a standard solution of 100µmol/mL was prepared by adding 667µL distilled water and stored at 2-8°C for 8 weeks.

Product Description:

Xylanase (EC 3.2.1.8), produced mainly by microorganisms, catalyzes the hydrolysis of xylan, also known as pentosanase or hemicellulase, can break down the cell wall of raw materials in the brewing or feed industry as well as beta-glucan, reduce the viscosity of brewing materials, promote the release of active substances, and reduce non-starch polysaccharides in feed. Promote the absorption and utilization of nutrients, so it is widely used in the brewing and feed industry, acid xylan (ACX) is generally isolated from acid-resistant fungi, bacteria and some molds.

ACX can degrade xylan into reducing oligosaccharides and monosaccharides in acidic environment, and further react with 3, 5-dinitrosalicylic acid in boiling water bath. There is a characteristic absorption peak at 540nm, and the color depth of the reaction solution is proportional to the amount of reducing sugar produced by enzymatic hydrolysis. The activity of ACX can be calculated by measuring the increase rate of absorption value of the reaction solution at 540nm.

Required but not provided:

Mortar/homogenizer, desk centrifuge, spectrophotometer, water bath, adjustable pipette, 1 mL glass cuvette, ice and distilled water.

Procedure:

I. Sample preparation

1. Fermentation solution: the fermentation solution is centrifuged at 8000rpm, at 4°C, for 15min, supernatant is taken and placed on ice for testing.

2. Dry enzyme powder: weigh 1mg, add 1mL Buffer Fluid, fully dissolve after shaking and put on ice to be measured.

3. Tissue: Weigh 0.1g tissue, add 1mL Buffer Fluid, fully grind on ice. 8000g, centrifuge at 4°C for 15min,

BC2600 - Page 1 / 4

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take supernatant, put on ice to be measured.

Note: Samples with high reducing sugar content (such as plant fruits, etc.) can be properly diluted with distilled water before determination.

II. Determination procedure

1. Preheat spectrophotometer for 30 minutes, adjust wavelength to 540 nm, set the counter to zero with distilled water.

2. dilution of standard solution: before the use of distilled water to dilute the standard product into
3, 2, 1, 0.5, 0.25, 0.125, 0.0625µmol/mL of standard solution to be measured.

3. quasi-dilution table:

Num	Predilution concentration	Standard	Buffer fluid	Diluted concentration
ber	(µmol/mL)	volume (μL)	volume (μL)	(µmol/mL)
1	100	100	900	10
2	10	150	350	3
3	10	100	400	2
4	10	100	900 💿	1
5	1	500	500	0.5
6	0.5	500	500	0.25
7	0.25	500	500	0.125
8	0.125	500	500	0.0625

Note: $200\mu L$ per tube is required in the experiment.

4. Sample determination (add the following reagents in 1.5 mL EP tube in turn).

Reagent Name (µL)	Contrast tube (A _C)	Test tube (A _T)	Blank tube (A _B)	Standard tube (A _S)
Sample	200	200		-
Standard	5	-	all hores -	200
Buffer fluid	300	300 90	500	300
Reagent I	-	200	200	200

Mix well, cap tightly, water bath at 50°C for 30min, immediately boil water bath for 10min to inactivate. (Be careful not to let the lid burst, so as not to water, change the reaction system)

Reagent I	200	-	5	-
Reagent II	300	300	300	300
Reagent III	100	100	100	100

Mix well, boiling water bath color development for 5min (be careful not to let the lid burst, so as not to change the reaction system), ice bath cooling as soon as possible to measure the absorption value at 540nm wavelength A_C , A_T , A_S , A_B , calculate $\Delta A_T = A_T - A_C$, $\Delta A_S = A_S - A_B$. Blank tubes and standard tube only need to be done 1-2 times.

III. Calculation

BC2600 - Page 2 / 4

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1. Drawing of standard curve:

The standard curve is established according to the concentration of the standard tube (x, μ mol/mL) and the absorbance ΔA_S (y, ΔA_S). According to the standard curve, the ΔA determination (y, ΔA_T) is

brought into the formula to calculate the sample concentration (x, μ mol/mL).

2. Fermentation fluid ACX activity calculation:

Enzyme activity definition: at 50°C and pH 4.8, the amount of enzyme required to decompose xylan to produce 1µmol reducing sugar per milliliter of fermentation solution per minute is one unit of activity of acid xylanase.

ACX activity $(U/mL) = x \div T \times F = x \div 30 \times F$

3. ACX activity calculation of dry enzyme powder:

Enzyme activity definition: under the condition of 50°C and pH 4.8, the amount of enzyme required to decompose xylan to produce 1µmol reducing sugar per milligram of enzyme per minute is one unit of activity of acid xylanase.

ACX activity (U/mg) = $x \times V_S \div (V_S \times W_E \div V_B) \div T \times F = x \div W_E \div 30 \times F$

4. Calculation of NEX activity in the organization:

(1) Calculated by sample protein concentration:

Enzyme activity definition: Under the condition of 50°C and pH 4.8, the amount of enzyme required for decomposing xylan to produce 1µmol reducing sugar per mg of hiprotein per minute is one unit of activity of acid xylanase.

ACX activity (U/mg prot) = $x \times V_S \div (V_S \times Cpr) \div T \times F = x \div Cpr \div 30 \times F$ (2) Calculated by sample quality:

Enzyme activity definition: Under the condition of 50°C and pH 4.8, the amount of enzyme required to decompose xylan to produce 1µmol reducing sugar per g tissue per minute is one unit of

activity of acid xylanase.

ACX activity (U/g mass) = $x \times V_S \div (V_S \times W_S \div V_B) \div T \times F = x \div W_S \div 30 \times F$

V_s: Added sample volume, 0.2mL;

V_B: Add Buffer Fluid volume, 1mL;

WE: Mass of enzyme dry powder, mg;

Ws: Sample mass, g;

Cpr: Protein concentration, mg/mL;

T: Reaction time, 30 minutes;

F: Sample dilution ratio;

Note:

The absorbance change should be controlled between 0.01 and 1.2, otherwise increase the

BC2600 - Page 3 / 4

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sample size

or dilute the sample, and pay attention to changing the dilution multiple in the calculation formula simultaneously.

Experimental example:

1. Take 0.1079g blueberry, add 1ml extract, grind it on ice. Take the supernatant and dilute it 40 times with distilled water and follow the determination procedure. Measured in a cuvette, and calculate $\Delta A_T = A_T - A_C = 0.717 - 0.581 = 0.136$, Bring the standard curve y=0.413x-0.0233 (R²=0.9993), calculate x=0.386, and calculate the ACX activity according to the sample mass:

ACX activity (U/g mass) = $x \div W \div 30 \times F = 0.386 \div 0.1079 \div 30 \times 40 = 4.770$ U/g mass.

2. Take pickle juice, centrifuge, take supernatant, and follow the determination procedure. Measured in a cuvette, and calculate $\Delta A_T = A_T - A_C = 0.639 - 0.392 = 0.247$, Bring the standard curve y=0.413x-0.0233 (R²=0.9993), calculate x=0.654, and calculate the ACX activity according to the fermentation fluid:

ACX activity $(U/mL) = x \div T \times F = 0.654 \div 30 \times 1 = 0.022$ U/mL.

Related Products:

BC2590/BC2595	Neutral xylanase (NEX) Activity Assay Kit
BC3610/BC3615	Alkaline xylanase (BAX) Activity Assay Kit



BC2600 – Page 4 / 4

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