

## Alpha-ketoglutaric acid ( $\alpha$ -KG) content Assay Kit

**Note:** Take two or three different samples for prediction before test.

**Operation Equipment:** Ultraviolet Spectrophotometer

**Catalog Number:** BC5420

**Size:** 50T/48S

### Components:

Extract I: Liquid 60 mL  $\times$ 1. Storage at 2-8°C.

Extract II: Liquid 9 mL  $\times$ 1. Storage at 2-8°C.

Reagent I: Liquid 40 mL  $\times$ 1. Storage at 2-8°C.

Reagent II: Liquid 4 mL  $\times$ 1. Storage at 2-8°C.

Reagent III: Powder  $\times$ 1. Storage at -20°C. Add 3.34 mL of distilled water to Reagent III and dissolve thoroughly before use. Unused reagents can be stored at -20°C for up to 4 weeks after dispensing, avoid repeated freezing and thawing.

Reagent IV: Powder  $\times$ 1. Storage at -20°C. Add 0.75 mL of distilled water to Reagent IV and dissolve thoroughly before use. Unused reagents can be stored at -20°C for up to 4 weeks after dispensing, avoid repeated freezing and thawing.

Standard: Powder  $\times$ 1. Storage at 2-8°C. Before use, add 0.856 mL of distilled water to the standard to dissolve it fully and prepare a standard solution of 80  $\mu$ mol/mL  $\alpha$ -ketoglutaric acid. The dissolved standard solution can be stored at 2-8°C for 4 weeks.

Reagent IV working solution: before use according to the sample amount according to reagent IV: distilled water = 0.1mL: 0.4mL (a total of 0.5mL, 10T) ratio of preparation, ready to use.

400nmol/mL Standard solution preparation: 50 $\mu$ L of 80 $\mu$ mol/mL standard solution taken before the experiment, add 950 $\mu$ L of distilled water and mix thoroughly to prepare 4 $\mu$ mol/mL standard solution. Then take 100 $\mu$ L of 4 $\mu$ mol/mL standard solution and 900 $\mu$ L of distilled water and mix to prepare 0.4  $\mu$ mol/mL (400nmol/mL) standard solution.

### Product Description

Alpha-ketoglutarate ( $\alpha$ -KG) is an important metabolic intermediate in the tricarboxylic acid cycle and a key link between intracellular carbon and nitrogen metabolism.  $\alpha$ -KG, as a short-chain carboxylic acid molecule, is a precursor to many important amino acids such as glutamine and glutamate, and is not only directly involved in energy supply, but also participates in a variety of intracellular chemical reactions and has a variety of physiological effects.

GDH catalyses the formation of glutamate and NAD<sup>+</sup> from NH<sub>4</sub><sup>+</sup>,  $\alpha$ -ketoglutarate and NADH, causing a decrease in absorbance at 340nm. The  $\alpha$ -ketoglutarate content was calculated by measuring the change in NADH.

### Reagents and Equipment Required but Not Provided.

Ultraviolet spectrophotometer, centrifuge, water bath/ heating dry baths / constant temperature incubator,

analytical balance, adjustable pipette, 1mL quartz cuvette, mortar/homogeniser/cell ultrasonicator, distilled water and ice.

## Procedure

### I. Sample preparation:

#### 1. Tissue:

Accordance the ratio of tissue(g) : extract I volume (mL)=1: 5~10 (add 1 mL of extract solution to 0.1 g of tissue), homogenate on ice. Centrifuge at 12000g for 10 minutes at 4°C. Take 0.8mL of supernatant, then slowly add 0.15mL of extract II, slowly blow and mix until no air bubbles are generated, centrifuge at 4°C for 10min at 12000g and then remove the supernatant for measurement.

#### 2. Bacteria or cells:

Accordance the ratio of cells amount( $10^6$ ) : extract I volume (mL)=5~10: 1 (add 1 mL of extract solution to 5 million cells). Ultrasonic on ice bath to smash cells, (powder 300w, ultrasonic 3s, interval 7s for 3 minutes). Centrifuge at 12000g for 10 minutes at 4°C. Take 0.8mL of supernatant, then slowly add 0.15mL of extract II, slowly blow and mix until no air bubbles are generated, centrifuge at 4°C for 10min at 12000g and then remove the supernatant for measurement.

#### 3. Liquid

Take 100 $\mu$ L of liquid and add 1mL of extract I, centrifuge at 4°C 12000g for 10min, take 0.8mL of supernatant, then slowly add 0.15mL of extract II, slowly blow and mix until no bubbles are produced, centrifuge at 12000g for 10min and take the supernatant for measurement.

### II. Determination Procedure

1. Preheat the ultraviolet spectrophotometer for more than 30 minutes, adjust the wavelength to 340 nm and set the counter to zero with distilled water.

2. Operation table: (Add the following reagents to a 1mL cuvette)

Reagent Name ( $\mu$ L)	Test tube(t)	Standard tube(s)	Blank tube(b)
Sample	300	-	-
Standard	-	300	-
Distilled water	-	-	300
Reagent I	550	550	550
Reagent II	50	50	50
Reagent III	50	50	50
Preheat at 37°C for 5min (water bath/ heating dry baths preheat recommended)			
Reagent IV working solution	50	50	50

Add reagent IV working solution and mix thoroughly immediately. The absorbance value A1 at 20s was measured at 340nm and quickly placed at 37°C for 5min. The absorbance value A2 at 5min20s was measured by wiping dry. Calculate  $\Delta A_t = A_{1t} - A_{2t}$ ,  $\Delta A_s = A_{1s} - A_{2s}$ ,  $\Delta A_b = A_{1b} - A_{2b}$ . Blank and standard tubes should only be measured 1-2 times.

### III. Calculation:

1. Calculation according to protein content

$$\begin{aligned} \alpha\text{-KG content (nmol/mg prot)} &= C_s \times (\Delta A_t - \Delta A_b) \div (\Delta A_s - \Delta A_b) \times V_s \div (C_{pr} \times V_s) \times F \\ &= 400 \times (\Delta A_t - \Delta A_b) \div (\Delta A_s - \Delta A_b) \div C_{pr} \times F \end{aligned}$$

2. Calculation according to quality of the sample

$$\begin{aligned} \alpha\text{-KG content (nmol/g weight)} &= C_s \times (\Delta A_t - \Delta A_b) \div (\Delta A_s - \Delta A_b) \times (V_{se} + V_{e2}) \div \\ &\quad (W \times V_{se} \div V_{e1}) \times F \\ &= 475 \times (\Delta A_t - \Delta A_b) \div (\Delta A_s - \Delta A_b) \div W \times F \end{aligned}$$

3. Calculation according to number of bacteria or cells

$$\begin{aligned} \alpha\text{-KG content (nmol}/10^6 \text{ cell)} &= C_s \times (\Delta A_t - \Delta A_b) \div (\Delta A_s - \Delta A_b) \times (V_{se} + V_{e2}) \div (N \times V_{se} \div V_{e1}) \times F \\ &= 475 \times (\Delta A_t - \Delta A_b) \div (\Delta A_s - \Delta A_b) \div N \times F \end{aligned}$$

4. Calculation according to volume of liquid

$$\begin{aligned} \alpha\text{-KG content (nmol/mL)} &= C_s \times (\Delta A_t - \Delta A_b) \div (\Delta A_s - \Delta A_b) \times (V_{se} + V_{e2}) \div [V_v \times V_{se} \div (V_v + \\ &\quad V_{e1})] \times F \\ &= 5225 \times (\Delta A_t - \Delta A_b) \div (\Delta A_s - \Delta A_b) \times F \end{aligned}$$

$C_s$ : Concentration of alpha-ketoglutaric acid standard solution, 400 nmol/mL;  $V_s$ : Volume of sample added to the reaction system, 0.3mL;  $V_{se}$ : Volume of supernatant at extraction, 0.8mL;  $V_{e2}$ : Volume of extract II added, 0.15mL;  $V_{e1}$ : Volume of extract I added, 1mL;  $V_v$ : Liquid sample volume, 0.1mL;  $C_{pr}$ : Protein concentration, mg/mL;  $W$ : Sample quality, g;  $N$ : Number of cells or bacteria,  $10^6$ ;  $F$ : Sample dilution.

#### Note:

1. If the sample  $A_{1t}$  is  $< A_{1 \text{ blank}}$  or  $\Delta A_t$  is  $> 0.8$ , dilute the sample with distilled water or shorten the second  $37^\circ\text{C}$  reaction time; if the  $\Delta A_t$  is  $< 0.01$ , increase the sample size or extend the second  $37^\circ\text{C}$  reaction time. Simultaneously modify the formulae for the final calculation.
2. Extract I contains a protein precipitant, therefore the supernatant cannot be used for protein concentration determination. If protein content is to be determined, a separate tissue should be taken.
3. Temperature has a significant effect on this experiment, so it is important to keep the reaction temperature at  $37^\circ\text{C}$ .

#### Experimental Examples:

1. 0.1066g of mouse liver tissue was weighed for sample processing, operated according to the assay procedure and measured with a 1mL quartz cuvette to calculate  $\Delta A_t = A_{1t} - A_{2t} = 1.201 - 1.175 = 0.026$ ,  $\Delta A_s = A_{1s} - A_{2s} = 1.015 - 0.487 = 0.528$ ,  $\Delta A_b = A_{1b} - A_{2b} = 1.080 - 1.074 = 0.006$ . Calculated by bringing into the equation:

$\alpha$ -KG content (nmol/g weight) =  $475 \times (\Delta A_t - \Delta A_b) \div (\Delta A_s - \Delta A_b) \div W \times F = 170.72$  nmol/g mass

**References:**

[1] Azmi N E, Ahmad M, Abdullah J, et al. Biosensor based on glutamate dehydrogenase immobilized in chitosan for the determination of ammonium in water samples[J]. Analytical Biochemistry, 2009,

388(1):28-32.

[2] Fei Ding, Qiannan Hu, Meiling Wang, et al. Knockout of SISBPASE Suppresses Carbon Assimilation and Alters Nitrogen Metabolism in Tomato Plants. International Journal of Molecular Sciences. December 2018;(IF4.183)

[3] Lin Y, Nan J, Shen J, et al. Canagliflozin impairs blood reperfusion of ischaemic lower limb partially by inhibiting the retention and paracrine function of bone marrow derived mesenchymal stem cells[J]. EBioMedicine, 2020, 52: 102637

**Related Products:**

- BC0080/BC0085 Nitrate Reductase(NR) Activity Assay Kit
- BC0710/BC0715  $\alpha$ -Ketoglutarate Dehydrogenase ( $\alpha$ -KGDH) Activity Assay Kit
- BC1460/BC1465 Glutamate dehydrogenase (GDH) Activity Assay Kit
- BC0980/BC0985 Acetyl Coenzyme A Content Assay Kit
- BC5490/BC5495 Malic acid content Assay Kit (WST colorimetric method)
- BC2150/BC2155 Citric acid(CA) Content Assay Kit