

# Soil Acid Xylanase (S-ACX) Activity Assay Kit

Note: Take two or three different samples for prediction before test.

**Detection equipment:** Spectrophotometer/Microplate reader

Cat No: BC5725

Size: 100T/48S

## **Components:**

**Buffer Fluid:** Liquid 35 mL×1, store at 2-8°C. **Reagent I:** Liquid 18 mL×1, store at 2-8°C.

**Reagent II:** Liquid 15 mL×1, store at 2-8°C.

**Standard:** Powder×1, 10mg xylose.

Standard dilution solution: Before use, prepare according to the ratio of buffer solution: Reagent I = 4mL: 2mL (6mL, about one standard curve dilution amount), and use it immediately.

Standard: Before use, a standard solution of  $100\mu$ mol/mL was prepared by adding  $667\mu$ L standard dilution solution and stored at 2-8°C for 8 weeks.

# **Product Description:**

Soil acid xylanase (S-ACX), also known as soil acidic hemicellulase, is mainly isolated from microorganisms with an optimal growth pH of 4-5.

In acidic environments, S-ACX catalyzes the degradation of xylan into reducing oligosaccharides and monosaccharides. Under boiling water bath conditions, it further undergoes a color reaction with 3,5-dinitrosalicylic acid, with a characteristic absorption peak at 540nm. The color depth of the reaction solution is directly proportional to the amount of reducing sugars produced by enzymatic hydrolysis. By measuring the rate of increase in absorbance of the reaction solution at 540nm, S-ACX activity can be calculated.

# **Required but not provided:**

Spectrophotometer/Microplate reader, micro glass cuvette/96 well plate, balance, desk centrifuge, water bath, 30-50 mesh sieve, distilled water.

# **Procedure:**

I. Sample preparation(The sample size to be tested can be adjusted appropriately, and the specific proportion can be referred to in the literature)

Fresh soil samples are naturally air dried or air dried in a 37 °C oven, and sieved through a 30-50 mesh sieve.

# **II. Determination procedure**

1. Preheat spectrophotometer/microplate reader for 30 minutes, adjust wavelength to 540 nm, set the counter to zero with distilled water.

2. Dilution of standard solution: Using **standard dilution solution** to dilute the Standard into 2, 1.5, 1.2, 1, 0.8, 0.4, 0.2µmol/mL of standard solution before measured.

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### 3. Quasi-dilution table:

Num -ber	Predilution concentration (µmol/mL)	Standard volume (µL)	Volume of <b>standard</b> dilution solution (µL)	Diluted concentration (µmol/mL)	
1	100	100	900	10	
2	10	200	800	2	
3	10	150	850	1.5	
4	10 4	120	880	1.2	
5	10	100	900	1	
6	<u>1</u>	200	50	0.8	
7	10 <sup>1</sup>	100	150	0.4	
8	ola tener	50	200	0.2	

Note:  $150\mu L$  per tube is required in the experiment.

4. Sample determination (add the following reagents in 1.5 mL EP tube in turn).

	Ň	<u> </u>		
Reagent Name (µL)	Control tube (A <sub>C</sub> )	Test tube (A <sub>T</sub> )	Blank tube (A <sub>B</sub> )	Standard tube (As)
Sample	0.05g	0.05g	1010 -	-
Buffer Fluid	200	200	C SOLEN	-
Reagent I	-	100	-	, iQ
Mix well and plac	e in a 50°C water bath		al a Chender	
Immediately after, heat	in a boiling water bath	0	SUFES	
deactivate (be careful	not to let the lid burst of	- (Y	-	
water from entering ar	nd altering the reaction s			
t	room temperature.	30		
Reagent I	100	-	alences -	-
At room temperature, o	centrifuge at 12000g for		0	
ta	ke the supernatant		-	1010
Supernatant	150	150	-	CO SOLENT
Standard	0.0	-	- 4	150
standard dilution	arpholes		150	e)
solution	SOLESOLE	-	150	-
Reagent II	100	100	100	100

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Mix well, accurately color in a boiling water bath for 5 minutes (be careful not to let the lid burst open to avoid water entering and altering the reaction system), cool to room temperature. Centrifuge at room temperature for 5 minutes at 12,000g. Take 200µL the supernatant and measure the absorbance at 540nm in 96 well plate or micro glass cuvette. Record the absorbance values as  $A_C$ ,  $A_T$ ,  $A_B$ ,  $A_S$ , calculate  $\Delta A_T = A_T - A_C$ ,  $\Delta A_S = A_S - A_B$ . The blank tube and standard curve only need to be measured 1-2 times. Each test tube should have a corresponding control tube.

#### **III.** Calculation

1. Drawing of standard curve:

The standard curve is established according to the concentration of the standard tube (x,  $\mu$ mol/mL) and the absorbance  $\Delta A_S$  (y,  $\Delta A_S$ ). According to the standard curve, the  $\Delta A_T$  (y,  $\Delta A_T$ ) is brought into the formula to calculate the sample concentration (x,  $\mu$ mol/mL).

2. Calculation of soil S-ACX activity:

Enzyme activity definition: An enzyme activity unit of an acid xylanase is defined as the amount of enzyme required to produce 1  $\mu$ mol of reducing sugar per gram of soil per hour by decomposing xylan at 50°C and pH 4.8.

S-ACX activity (U/g soil) =  $x \times V_{RV} \div W \div T \times F = 0.15 \times x \div W \times F$ 

V<sub>RV</sub>: Total reaction volume, 0.3mL;

W: Sample mass, g;

T: Reaction time, 2h;

F: Sample dilution ratio;

#### Note:

- 1. If  $\Delta A_T$  is less than 0.01, the sample size can be appropriately increased or the reaction time can be extended by 50 °C before measurement; If the  $\Delta A_T$  is greater than 1.5 or  $A_T$  is greater than 1.5, the supernatant can be diluted with distilled water before measurement, and attention should be paid to synchronously modifying the dilution factor in the calculation formula.
- 2. It is recommended to use a spiral tube to prevent the lid from bursting during the boiling water bath process and change the reaction system.

### **Experimental example:**

1. Take 0.05g of Wild mushroom soil, perform the first reaction according to the operation table for 2 hours, then dilute the supernatant twice with distilled water and follow the measurement steps. Measure with a 96 well plate, calculate  $\Delta A_T = A_T - A_C = 0.431 - 0.099 = 0.332$ , bring in the standard curve y=0.6582x-0.153 (R<sup>2</sup>=0.999), calculate x=0.737, and calculate S-ACX activity based on sample mass:

S-ACX activity (U/g soil) =  $0.15 \times x \div W \times F$  =4.422 U/g soil.

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